



## PLANTS FLAVONOIDS: CHEMISTRY, METABOLISM AND BIOLOGICAL ACTIVITIES

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### Abstract

Flavonoids are plant derived secondary metabolites. They are found in many fruits, vegetables, and seeds. They give the characteristic color, odor and taste. They are bioactive polyphenolic compounds and play a variety of roles in plants, including regulation of cell growth, attract insects and pollinators, and defending the biotic and abiotic stresses. Cell growth, attract insects and pollinators, and defending the biotic and abiotic stresses. Flavonoids gain attention in recent years due to high mortality rate of cardiovascular disease and low preventing rate of chronic cardiovascular disease. These compounds have anticancer, anti-aging, anti-inflammatory, neuroprotective cardio protective, immunomodulatory, anti-microbial, antidiabetic, anthelmintic, and antiviral effects in humans. In the review we discuss the current research on flavonoid, chemistry of flavonoids, their metabolism in human body, and their biological activities.

**Keywords:** Flavonoids, classification, biological activities, secondary metabolites

### Introduction

Flavonoids are secondary metabolites found in high concentrations in fruits, plants, and seeds and are gives for color, odor, and taste. Flavonoids play a variety of roles in plants, including regulation of cell growth, attract insects and pollinators, and defending the biotic and abiotic stresses [1]. These

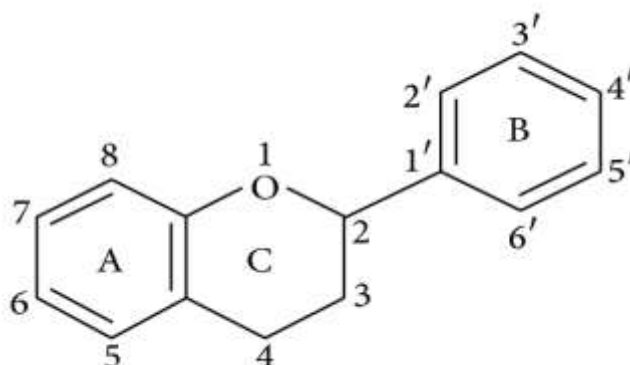
compounds have anticancer, anti-aging, anti-inflammatory, neuroprotective cardio protective, immunomodulatory, anti-microbial, antidiabetic, anthelmintic, and antiviral effects in humans [2]. Flavonoids have a C<sub>6</sub>-C<sub>3</sub>-C<sub>6</sub> flavone skeleton with two benzene rings connected by a three-carbon pyran ring C. Catechol B-ring position on the pyran C-ring, as well as the number and position of hydroxy groups on the catechol group of the B-ring, affect the antioxidant capacity of flavonoids [3]. Flavonoids are classified into several subgroups, including flavones, chalcones, isoflavonols, and flavonols. All the subgroups have their own sources. Onions and tea, for instance, are essential nutritional sources of flavones and flavonols. Flavonoids involve in protective role in a variety of biotic and abiotic stresses, as well as acting as unique UV filters [4].

Flavonoids are widely recognized as an essential compound having wide range of application in medicinal, nutraceutical, cosmetic, pharmaceutical industries. This is due to anti-oxidant, anti-mutagenic, inflammatory, anti-carcinogenic potential. They have an ability to alter the function of key enzymes. Flavonoids gain attention in recent years due to high mortality rate of cardiovascular disease and low preventing rate of chronic cardiovascular disease. The mechanisms of flavonoids are poorly known. Plant-derived derivatives have a wide range of biological activity. Emerging patterns in flavonoids research and development revolve around the extraction, isolation, characterization, identification and role of flavonoids in human health. Industries used Molecular docking and bioinformatics knowledge to determine the potential role of flavonoids. [5]. In the review we discuss the current research on flavonoid, chemistry of flavonoids, their metabolism in human body, and their biological activities.

### Chemistry of Flavonoids

Flavonoids are plant derived well-known bioactive Phenolic compounds. They were discovered in 1930. First flavonoid compound was isolated from orange. They were thought that it was new class of vitamin and gave the name vitamin P. Further research and studies concluded that they were not vitamin but flavonoid. Until now four thousand flavonoids have been isolated. [6].

Flavonoids have fifteen carbon chain in which two benzene ring present which are connected by heterocyclic pyrane ring. Flavonoids are classified in to different classes i.e. Flavones, flavonols, flavanones and others. The flavonoids classes are different on the base of carbon ring pattern and their mode of oxidation [6].



Flavonoids present as glycosides, aglycones and methylated derivatives. The aglycones is the basic flavonoid structure (Figure 1). Benzene ring condensed with six carbon ring either by  $\alpha$ -pyrone or by its dihydroderivative [7]. Flavonoids mostly hydroxylated at positions 3, 5, 7, 2, 3', 4', and 5'. Acetyl esters and Methyl ethers of alcohol group occur in nature. When glycosides formed, the glycosidic linkage is normally present at positions 3 or 7 and carbohydrate can be L-rhamnose, glucorhamnose, arabinose D-glucose or galactose [8].

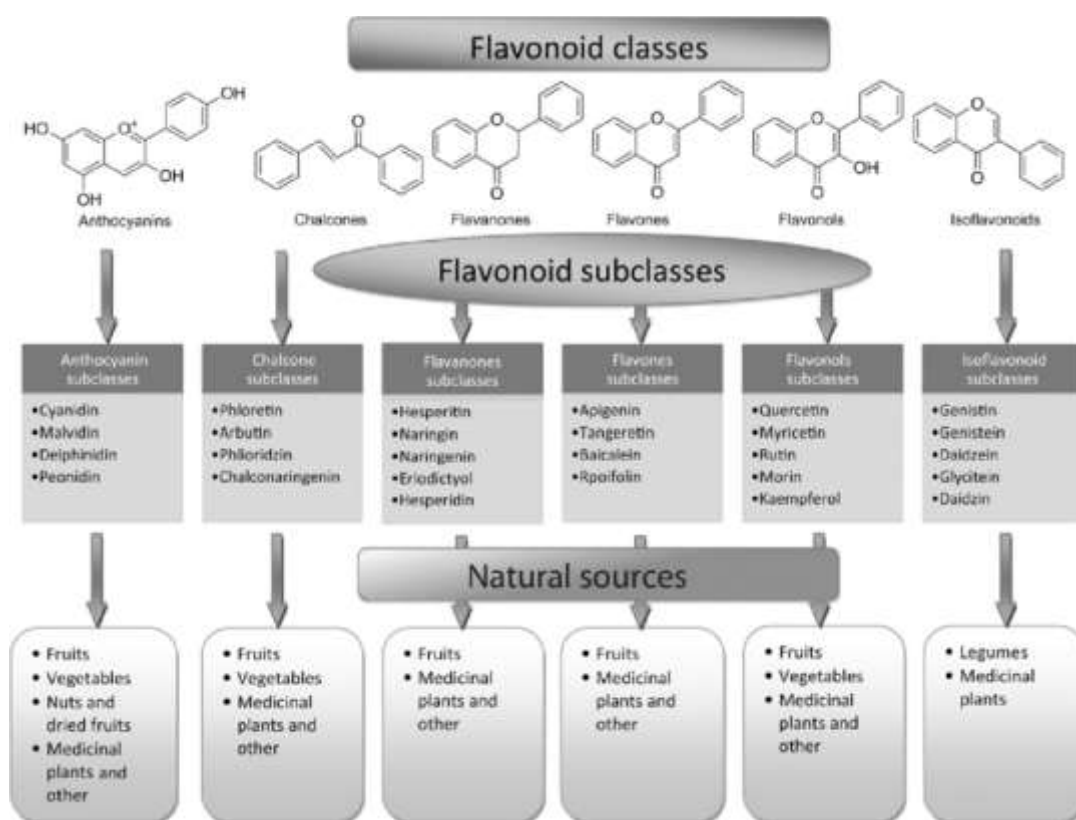
## Classification

Flavonoids are classified into different subgroups. The sub groups are differing on the bases of Carbon ring on which ring B attached. The groups are also differing on the base of unsaturation and C-ring oxidation. Following are the subgroups of flavonoids

1. Flavones
2. Flavonols
3. Flavanones
4. Flavanonols
5. Flavanols or catechins
6. Anthocyanins
7. Chalcones

## Flavones

Flavones are the subgroups of flavonoids. These are present generally in different parts of plants and richly present in parsley, Celery, peppers, mint, ginko biloba and chamomile. Apigenin, Luteolin, and tangeretin are flavones. Highest amount of polymethoxylated flavones, nobiletin, sinensetin and tangeretin present in citrus fruit. [9]. The hydroxyl group of flavones present in position 5 of ring A in vegetables and natural products. The hydroxylation in different conditions, generally in place seven of A ring or 3' and 4' of the B ring, may shift as indicated by ordered characterization of specific vegetable or organic product.



[6]

## Flavonols

Keto group is present in these flavonoids. Flavonols are building blocks of proanthocyanins. Many fruits and vegetables contains flavonoids. Extensive research has been done on some flavonoids such as quercetin, liquiritin, kaempferol and myricetin (Fig. 2). Tomato, berries apples, onions, lettuce and grapes are rich source of flavonoids. Leafy food such as tea, lemon grass and green tea are also rich source of flavonoids. They have many therapeutic activities. Flavonols involves in curing vascular infections. They are strong antioxidant. Flavonol are the largest class of flavonoids and are present in

most of fruits and vegetables. Quercetin is one such example which of largely distributed flavonol [10].

### **Flavanones**

Flavanones are mostly present in citrus fruits like lemon, grapes and oranges. Naringenin, and Hesperitin, are flavanones. Flavanone have antioxidant potential that's why they have number of therapeutic effects. Citrus flavonoids have many pharmacologic al potentials. They are also considering as bitter principle that's why citrus fruit peel is bitter in taste. They are antioxidant, antihyperlipidimic and anti-inflammatory. Flavanones, are also known as dihydroflavones. Saturation is present between double bond present at position two and three. (10).

### **Isoflavonoids**

Isoflavonoids are an enormous and extremely particular subclass of flavonoids. Iso flavonoids present in limited number of plants. These are commonly found in leguminous plants such as soyabeans. Some isoflavonoids are also present in microbes(11). They act as precursors forphytoalexins(12,13). Isoflavonoids have many therapeutic efficacies. Isoflavones, for example, daidzein and genistein are generally considered as phyto-oestrogens. Their oestrogenic effect is also determined (Fig. 2) (14).

### **Neoflavonoids**

Neoflavonoids are polyphenolic compounds. The neoflavone isolated from the first time from seeds of *Calophyllum inophyllum*. This neoflavonoids was callophylloides. It is also found in the bark of *Mesua thwaitesii* [15,16,17].

### **Flavanols**

Flavanols also known as flavan-3-ols or catechinsare 3-hydroxy derivatives of flavanones. These are multisubstituted and diverse class. In this class hydroxyl group is present on position 3 of carbon ring. Therefore, these are also known as flavon-3-ols. In this class no double bond is present between 2 and 3 positions. Flavones are found richly in apples, bananas, blueberries, pears and peaches.

### **Anthocyanins**

Anthocyanins are coloring agents in plants, fruits and flowers. Cyanidin, malvidin, delphinidin, peonidin and pelargonidin are common anthrocyanins (Fig. 2). They are present certain fruits like cranberries, blue berries, red grapes,dark currants, strawberries and raspberries. These compounds are utilizedin food industries in several applications [18]. Anthocyaninscolor depends upon the pH and methylationat the hydroxyl group on the A and B rings [10].

### **Chalcones**

Chalcones are open chain flavonoids. In these flavonoids C ring is absent in basic structure. Some examples of isolated chalcones are arbutin, phloretin, chalconaringenin and Phloridzin. Chalcones abundantly found in in pears, tomatoes, bear berries and wheat. Chalcones and their drivatives have many nutritional and medicinal potential [19-20]. The intake of naturally occurring chalcones have significant role in combating diseases.

### **Flavonoids Metabolism**

The dietary flavonoids absorption depends on their size of molecules, lipophilicity, configuration, solubility and pKa. Absorption of dietary flavonoids takes place in small intestine. Sometime before absorption the enter in the colon. The absorption may also depend on the flavonoids structure, whether it is aglycone or glycosides. Small intestine easily absorbed aglycan, whereas, glycoside for absorption first convert in to aglycon form. Flavonoids mostly bound with sugar *b*-glycosides in plants except sub class catechin [21].The quercetin is hydrophilic glucosides. Quercetin transports across

the intestine by Na<sup>+</sup>-dependent glucose cotransporter (SGLT1) [21]. Lactase phloridzin hydrolase is an enzyme present in the intestinal brush border. This enzyme hydrolyzes flavonoid glucosides that result in the liberation of aglycone which is then absorbed by the small intestine [22]. Lactase phloridzin hydrolase enzyme substrate specificity is not the same for every flavonoid [23]. Therefore, the glycosides which are not hydrolyzed by these enzymes are transported toward the colon where intestinal bacteria have the ability to hydrolyze flavonoid glycosides, but they will also degrade the liberated aglycones [24]. Since in the colon absorption capacity is compared to the small intestine. After absorption from the small intestine, the flavonoids are conjugated in the liver by sulfation, or methylation and converted into smaller phenolic compounds [25]. Because of conjugation reactions, the plasma and urine are free of flavonoid aglycones. [26]. Flavonoids from different food sources are different. They have different rates of absorption. For instance, quercetin from onion is absorbed more quickly than quercetin from tea and apple [27]. The flavonoids which are not absorbed by the small intestine move towards the colon where they are degraded by the microflora of the small intestine. The microflora breakdown the ring structure of flavonoids. In the stomach's acidic environment, large oligomers are hydrolyzed into monomers. Among all subclasses of flavonoids, isoflavones exhibit the highest bioavailability [28]. The flavonoids in green tea, after ingestion, are rapidly absorbed and enter the systemic circulation, causing a significant increase in plasma antioxidant levels [29].

### **Extraction, Isolation, and Characterization of Flavonoids**

Flavonoid solubility varies in different solvents. The solvent is selected on the basis of polarity of flavonoids. For example, ethyl acetate is used for aglycone extraction. The solvents that are toxic, such as chloroform and benzene, are not used in extraction. The polar aglycones are extracted with alcohol, acetone, and water [30,31]. The characterization of flavonoids is done by mass spectroscopy and nuclear magnetic resonance. However, some other techniques are also used for characterization, such as ultra-violet absorption spectroscopy and infrared spectroscopy [32].

Several researches have been published on the importance of analysis with MS fragmentation [33,34]. They highlight some main fragments. Most of which cleave the heterocyclic ring and are able to identify the presence of substituents in the flavonoid nucleus [35, 36,37].

The evaluation of structure is most commonly determined by nuclear magnetic resonance. Different groups can be determined due to the presence of the heterocyclic ring [38]. For instance, isoflavones differ from flavones on the basis of H-2 and C-2 chemical shifts [32].

### **Biological Potential of Flavonoids**

#### **Antibacterial Activity**

Flavonoids inhibit the biofilm formation of *Aeromonas hydrophila*, *Streptococcus mutans*, and *Escherichia coli*. [37] Quercetin and kampferol inhibit the formation of biofilm in *E. coli* [38]. Presence of hydroxyl phenolic groups in flavonoids is responsible for antibacterial effects. This group inhibits enzymes of bacteria and also interferes in their synthesis [39].

#### **Antifungal activity**

Many flavonoids have antifungal activity [40]. Flavonoids inhibit the growth of pathogenic fungal spores in plants and thus prevent fungal diseases in humans [41]. Flavonoids form complexes with protein in fungal cell walls. Prenylated flavones from the Moraceae family have antifungal activity against *Saccharomyces cerevisiae* and *Candida albicans* [40]. Baicalein, a flavone, is isolated from *Scutellaria baicalensis* and exhibits anticandidal activity against *Candida albicans*, *Candida parapsilosis*, and *Candida tropicalis* [42].

#### **Antiviral Activity**

Flavonoids exhibit anti-viral activity. They inhibit the enzymes involved in the virus life cycle. Flavon-3-ol inhibits HIV-1 and HIV-2 and other immunodeficiency viruses [43]. Anti-dengue activity of certain flavonoids such as naringin, hesperetin, and quercetin has been seen [44].

### **Steroid-genesis modulators**

Abyssinones flavonoids are steroid-genesis modulators against the enzymes aromatase, 3 $\beta$ -hydroxysteroid dehydrogenase, 17 $\beta$ -hydroxysteroid dehydrogenase of the steroid-genesis pathway [45].

### **Xanthine oxidase modulators**

Xanthine oxidase involves in gouty arthritis [46, 47, 48]. Flavonoids such as Licoisoflavone-A extracted from root of *Glycyrrhiza glabra* showed inhibitory effect against xanthine oxidase [49]. The flavonoidstricin, fisetin, butein, diosmetin, genistein, tricetin, vitexycarpin, rhamnetin, herbacetin, isorhamnetin, robinetin, biochanin, peonidin and okanin showed inhibitory activity. Benzopyran ring is present in flavonoids basic structure and involves in XO-inhibitory activity. [50]. Lin *et al.* studied many flavonoids having xanthine oxidase inhibitory activity [51]. Luteolin and Apigenin 5-hydroxy-6, 7, 3', 4'-tetramethoxyflavone showed inhibitory potential of *gnaphalium affine* against xanthine oxidase activity [52].

### **Anti-cholinesterase activity**

Acetylcholinesterase enzyme present in central nervous system and inhibit neural acetylcholine levels which is involve in the relives of Alzheimer disease [53]. Several number of flavonoids such as macluraxanthone and quercetin showed inhibitory activity against AChE and cholinesterase [54]. macluraxanthone exhibit strong inhibitory activity [55].

### **Anti-oxidant Activity**

Studies showed anti-oxidant activity of flavonoids [56]. Antioxidant are chemical compounds and protect the human from damage to reactive oxygen species. Flavonoids are the phytochemicals exhibit antioxidant activity and thus inhibits the disease causing agents [57-58].

### **Antiaging property**

Since many factors involve in aging process, but the most common cause is the oxidative damage of the cells [59] [60]. Study showed polyphenol present in açai pulp have antiaging properties because they have the ability to remove superoxide and peroxy radicals [61]. Quercetin and resveratrol flavonoids causes antiaging effects [62].

### **Anticancer Activity**

Many flavonoids isolated from *Achyrocline satureioides* have been shown to possess anticancer activity. The flavonoids isolated from this plant are luteolin, quercetin, 3-O-methyl-quercetin and achyrobichalcone [63].

### **Future research and development programs**

Flavonoids gained more attention in research since past ten years. They have many beneficial effects. Number of *in vitro* and *in silico* studies have been done to determine the therapeutic role of flavonoids. There is a need of further studies on to evaluate the health related benefits of flavonoids. Flavonoids study is complex as heterogeneity present in different molecular structures and less data available on the bioavailability. Moreover, deficient methods are available to determine the *in vivo* oxidative damage. There is need of techniques to determine the absorption and excretion of flavonoids. No data is available on the consequences of the chronic ingestion of flavonoids. *In silico* and molecular docking studies are needed to evaluate the molecules of flavonoid and their therapeutic efficacy in human beings. More studies are needed to search the more and more flavonoids so that the harmful synthetic medicine are replace by new molecules.

### **Conclusion**

Flavonoids are naturally occurring bioactive compounds and have number of beneficial effects. They are plants secondary metabolites. Studies reported that flavonoids have antioxidant, hypoglycemic,

anticancer, cardio protective, ant parasitic, antiviral, antiaging and antimicrobial activities. Flavonoids now recognized as an essential component in a wide range of nutraceutical, pharmaceutical, medicinal, and cosmetic applications. The mechanisms of flavonoids are poorly known. Emerging patterns in flavonoids research and development revolve around the isolation, identification, characterization, and role of flavonoids to cure disease. Molecular docking and bioinformatics knowledge are also being used by industry to estimate potential applications and manufacturing.

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